

Multielectron atoms, and bonding in H₂⁺

Note Title

12-11-2009

Final exam

Syllabus: Chemical potential, chemical and phase equilibrium, and all of quantum chemistry

Weightage: ~30% for thermo and rest quantum

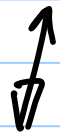
He atom

$$\psi_{1s}(1) \psi_{1s}(2) - 9.5$$

$$- \psi_{1s}(1) \psi_{ne}(2)$$

$$\psi_{n,l}(1) \psi_{A_2 l_2}(2)$$

exchange
degeneracy



$2s$

$$- \psi_{1s}(2) \psi_{ne}(1)$$

$$a = \pm b$$

$$\psi = a \psi_{1s}(1) \psi_{ne}(2) + b \psi_{1s}(2) \psi_{ne}(1)$$

Any property is independent of
exchange of particle labels

$$a = \pm b$$

$$\Psi = \begin{cases} \text{singlet} \\ \underline{\Psi_+} = \frac{1}{\sqrt{2}} (\Psi_{1s}(1) \Psi_{n\ell}(2) + \Psi_{1s}(2) \Psi_{n\ell}(1)) \\ \text{triplet} \\ \underline{\Psi_-} = \frac{1}{\sqrt{2}} (\Psi_{1s}(1) \Psi_{n\ell}(2) - \Psi_{1s}(2) \Psi_{n\ell}(1)) \end{cases}$$

If $r_1 = r_2$

$\Psi_- = 0 \Rightarrow$ electrons are anti-correlated

$$\langle \hat{H} \rangle = \text{Energy} \quad \hat{H} = \frac{-\hbar^2}{2m} \nabla_1^2 - \frac{\hbar^2}{2m} \nabla_2^2 - \frac{Ze^2}{r_1} - \frac{Ze^2}{r_2} + \frac{e^2}{r_{12}}$$

$$\langle E \rangle = \frac{\int \psi^* \hat{H} \psi \, dz_1 \, dz_2}{\int \psi^* \psi \, dz_1 \, dz_2}$$

$$\psi = \frac{1}{\sqrt{2}} (\psi_{1s}(1) \psi_{2s}(2) + \psi_{1s}(2) \psi_{2s}(1))$$

$$\text{Denominator} = \frac{1}{\sqrt{2}} \times \frac{1}{\sqrt{2}} \int (\psi_{1s}^*(1) \psi_{2s}^*(2) + \psi_{1s}^*(2) \psi_{2s}^*(1))$$

$$= \frac{1}{2} \left\{ \int \underbrace{\psi_{1s}^*(1) \psi_{1s}(1)}_{1} \underbrace{\psi_{2s}^*(2) \psi_{2s}(2)}_{1} \, dz_1 \, dz_2 + \dots \right.$$

$$\int \frac{1}{\sqrt{2}} (\psi_{1s}(1) \psi_{2s}(2) + \psi_{1s}(2) \psi_{2s}(1)) \left\{ -\frac{\hbar^2}{2m} \nabla_1^2 - \frac{\hbar^2}{2m} \nabla_2^2 - \frac{2}{r_1} - \frac{2}{r_2} + \frac{e^2}{r_{12}} \right\}$$

$$\frac{1}{2} \int \psi_{1s}(1) \left(-\frac{\hbar^2}{2m} \nabla_1^2 \psi_{1s}(1) \right) d\tau_1 \times \int \psi_{2s}(2) \psi_{2s}(2) d\tau_2$$

$$\int \frac{1}{\sqrt{2}} \left(\psi_{1s}(1) \psi_{2s}(2) \frac{e^2}{r_1} \psi_{1s}(1) \right) \psi_{2s}(2) d\tau_1 d\tau_2$$

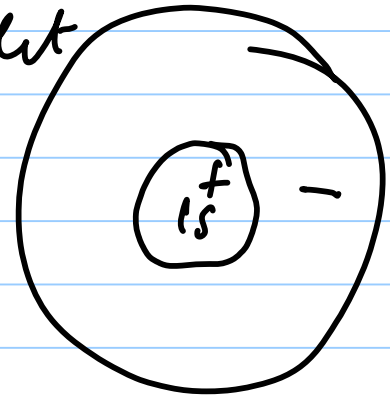
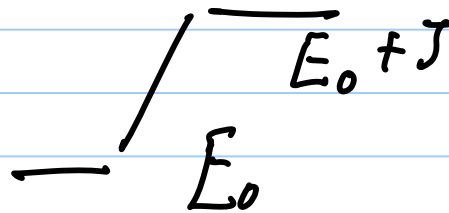
Con Lamb integral $\neq \int e \psi_{1s}(1) \psi_{1s}(1) \frac{1}{r_1} e \psi_{2s}(2) \psi_{2s}(2) d\tau_1 d\tau_2$

$$J = \int \underbrace{\psi_{1s}(1) \psi_{1s}(1)}_{f(1)} \underbrace{\frac{e^2}{r_{12}}}_{\text{inter-electronic repulsion}} \underbrace{\psi_{2s}(2) \psi_{2s}(2)}_{f(2)} d\tau_1 d\tau_2$$

exchange
integral
 $K =$

$$\int \psi_{1s}(1) \psi_{1s}(2) \frac{e^2}{r_{12}} \psi_{2s}(1) \psi_{2s}(2) d\tau_1 d\tau_2$$

$E_0 + J + K$ → singlet
 $E_0 + J - K$ → triplet (positive)



Multi electron atom (general case)

$$\hat{H} = \sum_{i=1}^N \underbrace{-\frac{\hbar^2}{2m_i}}_{K_i} \nabla_i^2$$

$$\left[\frac{Ze^2}{r_i} + \sum_{i < j} \frac{e^2}{r_{ij}} \right]$$

↓ nuclear attraction
↓ interelectronic repulsion

$$\hat{H} = \sum_i \hat{H}_i$$

$$\hat{H}_i = \frac{\hbar^2}{2m_i} \nabla_i^2 + V_{CF}(r_i)$$

$V_{CF}(r_i)$
 central field

$$E = E_1 + E_2 + \dots$$

$$\psi = \psi_1 \psi_2 \dots$$

Evaluate

$$V_{CF}(r_i) = ?$$

$$C \quad 1s^2 \quad 2s^2 \quad 2p^2$$

Hartree self consistent field method

$$\psi_{n_1 l_1}(1) \quad \psi_{n_2 l_2}(2)$$

$$\psi_{1s}^{guess}$$

$$\psi_{2s}^{guess}$$

$$\psi_{2p}^{guess}$$

"self-consistency"

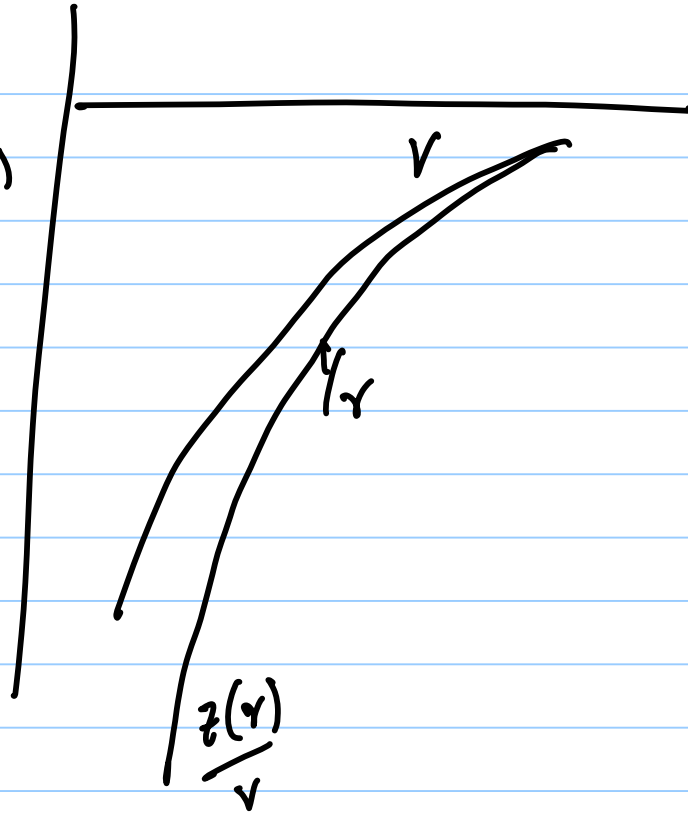
$$V_{eff}(1) = \int \frac{\psi_{n_2 l_2}^*(2) \psi_{n_2 l_2}(2)}{r_{12}} d\tau_2$$

$$\mathcal{H}_i \psi_i = \epsilon_i \psi_i$$

- 1) Guess some ψ based on intuition
- 2) Determine $V_{CF}(r_i)$
- 3) $\mathcal{H}\psi = E\psi$ and determine new ψ

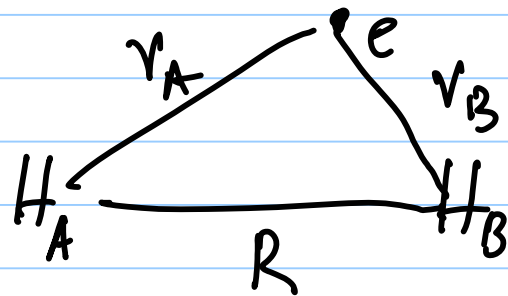
$Z(r)$ is a function of r

E becomes a function of r



Chemical bonding H_2^+

hybridization ← Valence bond theory - Heitler & London
 Molecular Orbital theory - Mulliken & Hund



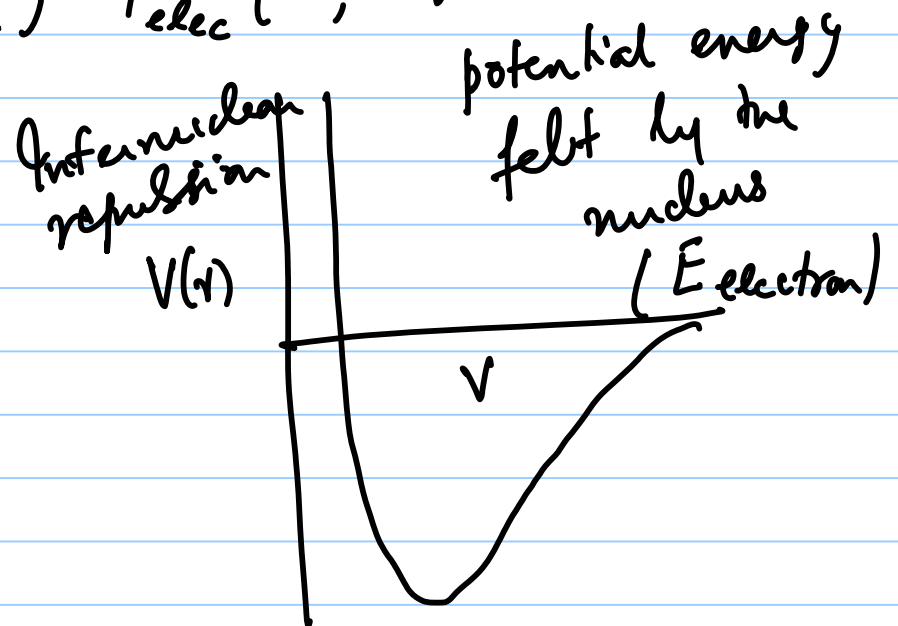
$$\hat{H} = \left\{ -\frac{\hbar^2}{2m_A} \nabla_A^2 - \frac{\hbar^2}{2m_B} \nabla_B^2 \right\} \left\{ -\frac{\hbar^2}{2m_e} \nabla^2 - \frac{e^2}{r_A} - \frac{e^2}{r_B} + \frac{e^2}{R} \right\}$$

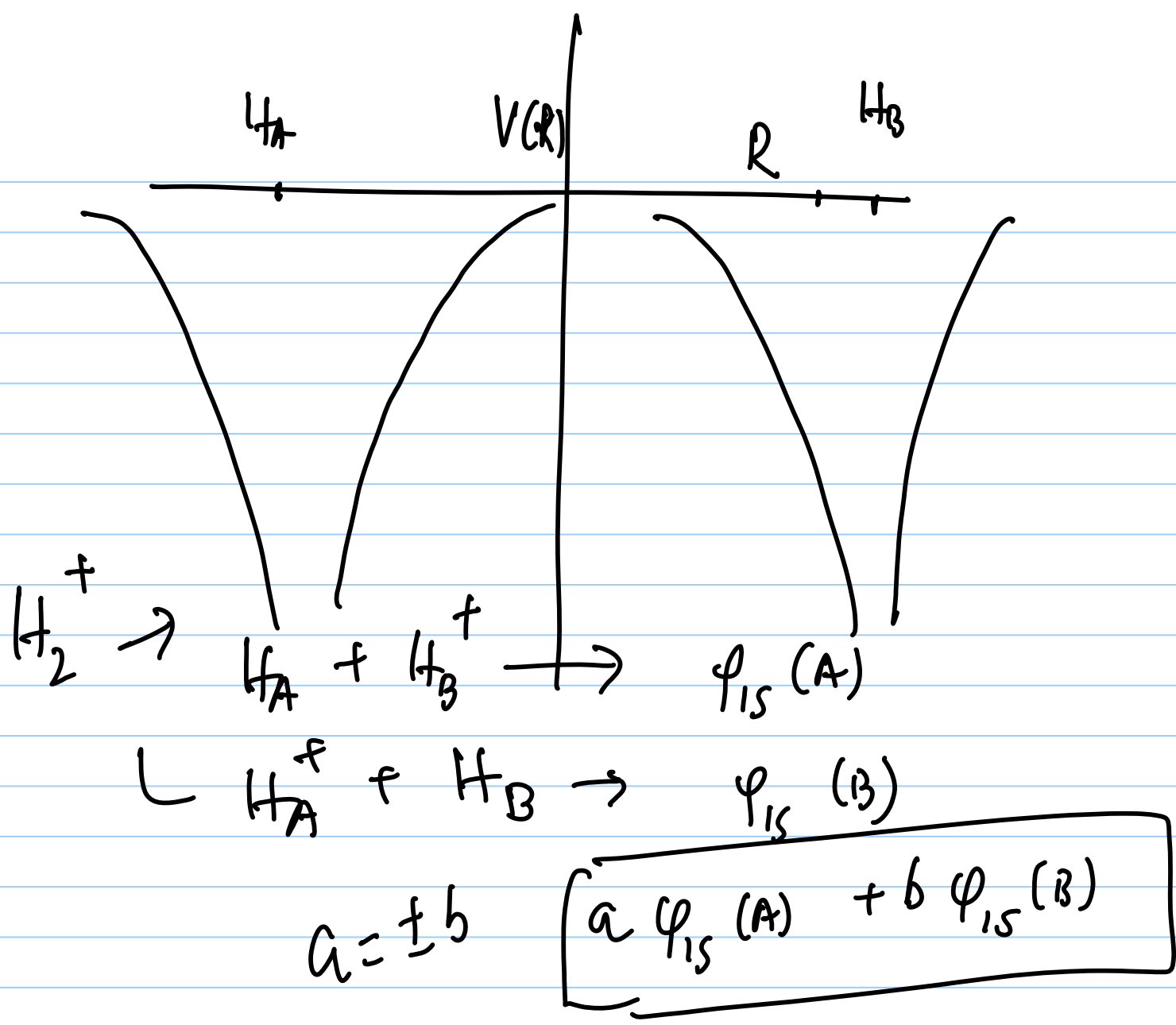
nuclear kinetic energy Heleee
 nuclear Ginter nuclear repulsion

$$\hat{H} = \hat{H}_{\text{nuc}} + \hat{H}_{\text{elec}}$$

Nucleus fixed (R) - Born - Oppenheimer approximation

$$\Psi = \Psi_{\text{nuc}}(R) \Psi_{\text{elec}}(r; R)$$





$$\Psi = a (\Psi_{1s}(A) \pm \Psi_{1s}(B))$$

$$\langle E \rangle_{R \rightarrow \infty} = \frac{\int \Psi^* \mathcal{H} \Psi d\tau}{\int \Psi^* \Psi d\tau}$$

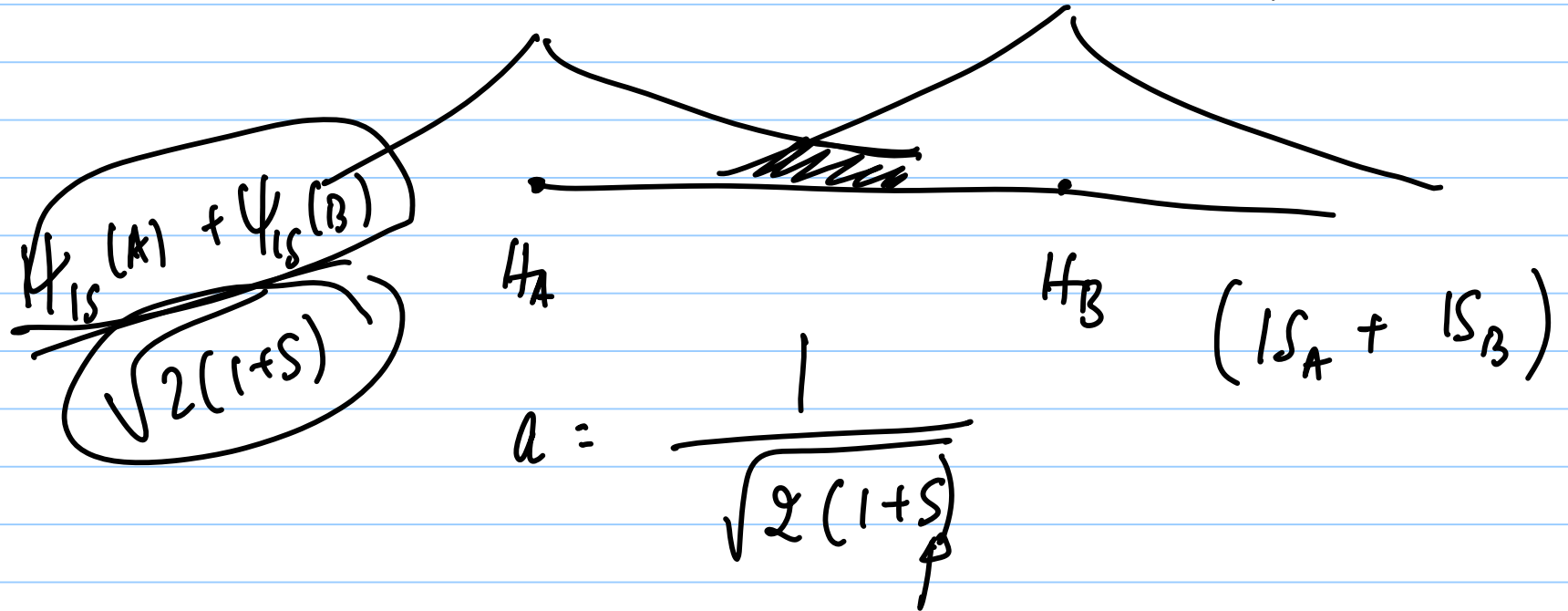
$$\text{Denominator} = \int \Psi^* \Psi d\tau$$

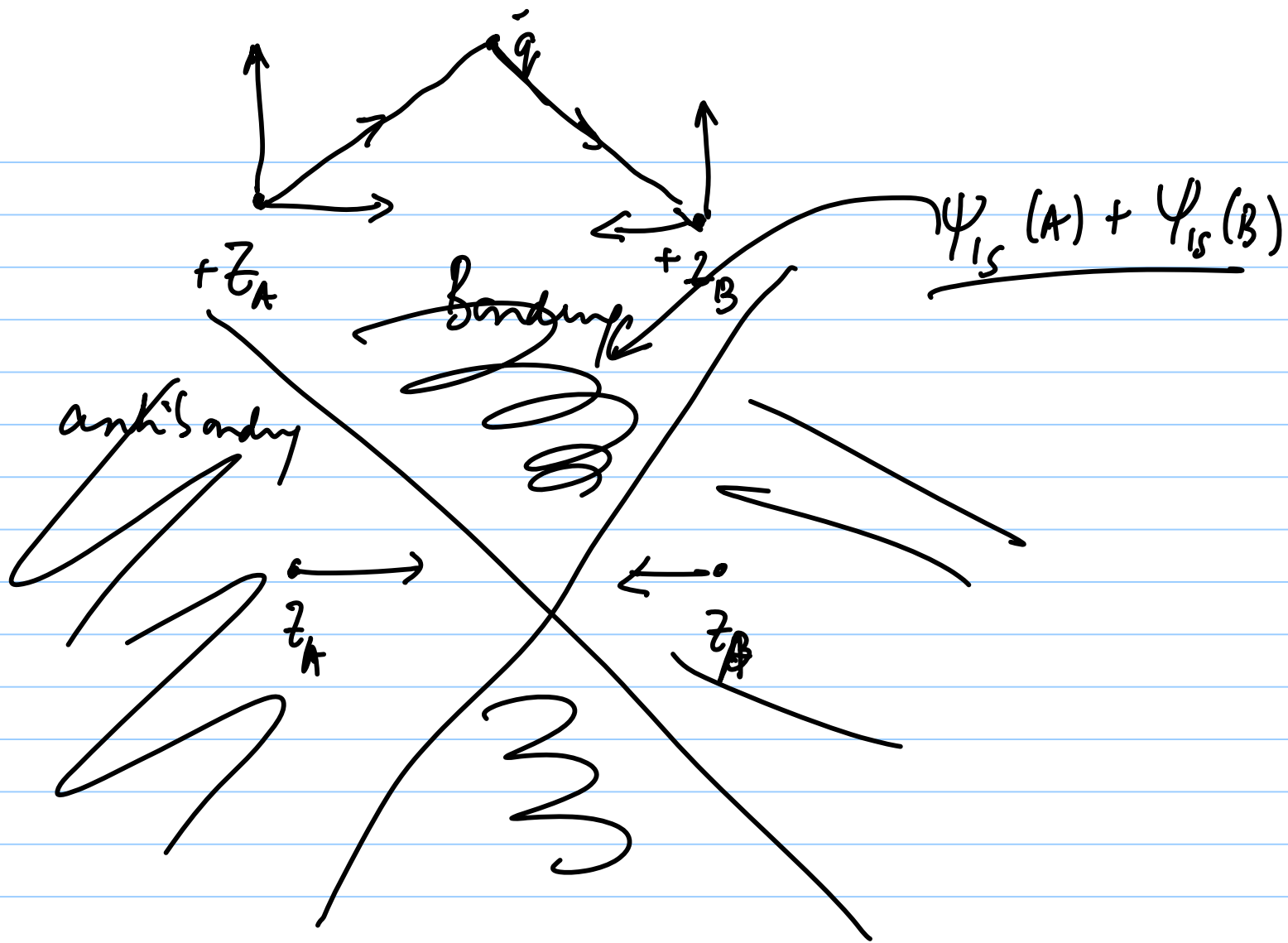
$$= a^2 \int (\Psi_{1s}(A) \pm \Psi_{1s}(B)) (\Psi_{1s}(A) \pm \Psi_{1s}(B)) d\tau$$

$$= a^2 \left\{ \int \Psi_{1s}(A) \Psi_{1s}(A) d\tau + \int \Psi_{1s}(A) \Psi_{1s}(B) d\tau + \int \Psi_{1s}(B) \Psi_{1s}(A) d\tau + \int \Psi_{1s}(B) \Psi_{1s}(B) d\tau \right\}$$

$$a = \frac{1}{\sqrt{2}}$$

$$S = \int \psi(A) \psi(B) dz \quad \text{overlap of the functions}$$





$$\langle \hat{E} \rangle = \int \frac{1}{\sqrt{2(1+S)}} (1S_A + 1S_B) \left(-\frac{1}{2} \nabla^2 - \frac{1}{r_A} - \frac{1}{r_B} + \frac{1}{R} \right)$$

H-atom

$$\hat{H} = \left(-\frac{\hbar^2}{2m} \nabla^2 - \frac{e^2}{r} \right)$$

$$\left(-\frac{1}{2} \nabla^2 - \frac{1}{r} \right) \int 1S_A \left(-\frac{1}{2} \nabla^2 - \frac{1}{r_A} \right) 1S_A + E_{H1S(A)}$$

$$\int (1S_A \left(-\frac{1}{2} \nabla^2 \right) 1S_B)$$

$$\int 1S_A \left(-\frac{1}{r_B} \right) 1S_A + \int 1S_A \frac{1}{R} 1S_A$$

$$+ \int 1S_A \left(-\frac{1}{r_A} \right) 1S_B + \int 1S_A \frac{1}{R} 1S_B$$