

Angular momentum and rotation

$$\vec{L} = \vec{r} \times \vec{p} = \begin{vmatrix} i & j & k \\ x & y & z \\ p_x & p_y & p_z \end{vmatrix}$$

$$\hat{L}_x = (\hat{y}\hat{p}_z - \hat{z}\hat{p}_y)$$

$$\hat{L}_x = \left(y \left(-i\hbar \frac{\partial}{\partial z} \right) - z \left(-i\hbar \frac{\partial}{\partial y} \right) \right)$$

Commutator relations

$$[\hat{L}_x, \hat{L}_y] = i\hbar L_z, \quad [\hat{L}_y, \hat{L}_z] = i\hbar L_x, \quad [\hat{L}_z, \hat{L}_x] = i\hbar L_y$$

$$[\hat{L}^2, \hat{L}_x] = [\hat{L}^2, \hat{L}_y] = [\hat{L}^2, \hat{L}_z] = 0$$

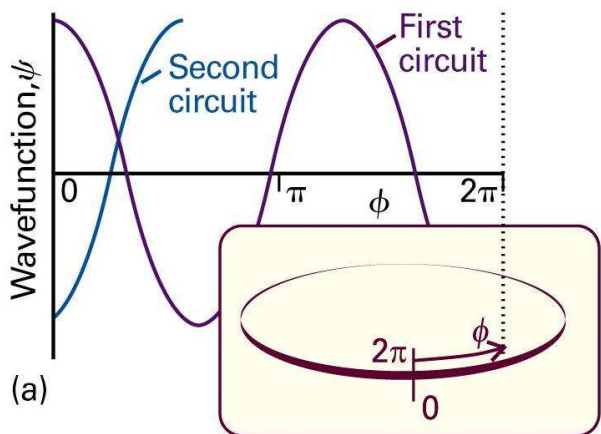
Definition of L^2 in spherical coordinates

$$\hat{L}^2 = -\hbar^2 \left(\frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \sin \theta \frac{\partial}{\partial \theta} + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \phi^2} \right)$$

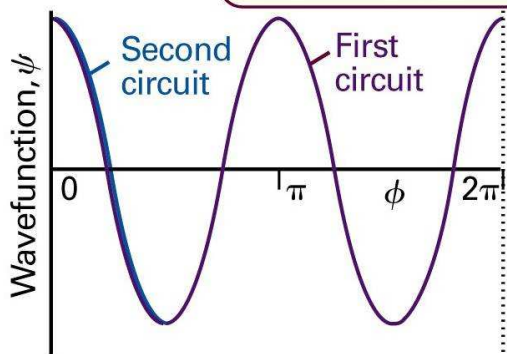
Rigid rotor confined to a plane

$$\hat{H} = \frac{\hat{L}^2}{2I} = -\frac{\hbar^2}{2I} \frac{\partial^2}{\partial \phi^2}$$

$$\hat{H} \psi = E \psi$$



(a)



(b)

$$\psi = A e^{im\phi} \quad m = \sqrt{\frac{2IE}{\hbar^2}}$$

$$\psi(\phi) = \psi(\phi + 2\pi)$$

Restricts

$$m = 0, \pm 1, \pm 2$$

Figure 9-28
Atkins Physical Chemistry, Eighth Edition
© 2006 Peter Atkins and Julio de Paula

$$E = \frac{m^2 \hbar^2}{2I} \quad m = 0, \pm 1, \pm 2, \dots$$

—	$\frac{\hbar^2}{2I}$
— ± 1	$\frac{4\hbar^2}{2I}$
— 0	$\frac{\hbar^2}{2I}$

$$L^2 = -\hbar^2 \left[\frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \sin\theta \frac{\partial}{\partial\theta} - \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial\varphi^2} \right]$$

Power series solution

$$L^2 \psi = k \psi \quad L^2(L_z \psi) = L_z L^2 \psi$$

$$L_z \psi = c \psi \quad \text{"} \quad L_z k \psi$$

Simultaneous eigenfunctions = $k(L_z \psi)$
of L_z and L^2

Rotor confined to the plane
 $L^2 = L_z^2 = -\hbar^2 \frac{d^2}{d\varphi^2}$

$$\bar{\Phi}(\varphi) = A e^{im\varphi} \quad A = \frac{1}{\sqrt{2\pi}}$$

$$\Psi(\theta, \varphi) = \Theta(\theta) \bar{\Phi}(\varphi)$$

$$L_z^2 \bar{\Phi}(\varphi) = -\hbar^2 \frac{d^2}{d\varphi^2} A e^{im\varphi}$$

$$L^2 = -\hbar^2 \left[\frac{1}{\sin\theta} \frac{\partial}{\partial\theta} \sin\theta \frac{\partial}{\partial\theta} + \frac{1}{\sin^2\theta} \frac{\partial^2}{\partial\varphi^2} \right] \Psi$$

$$L^2 \Psi = \lambda \hbar^2 \Psi$$

"Associated Legendre equation"

- a. $P(\theta) = \text{constant}$ $\psi(x, y) = \psi_x \psi_y$
 $\lambda = 0, m = 0$ $\psi(\theta, \varphi) = P(\theta) \Phi(\varphi)$
- b. $P(\theta) = \sin \theta$
 $\lambda = 2, m = \pm 1$
- c. $P(\theta) = \cos \theta$
 $\lambda = 2, m = 0$
- d. $P(\theta) = \cos^2 \theta - \frac{1}{3}$
 $\lambda = 6, m = 0$

$$\begin{array}{l} \lambda = 0, \quad m = 0 \\ \lambda = 2, \quad m = 0, \pm 1 \\ \lambda = 6, \quad m = 0, \pm 1, \pm 2 \end{array}$$

$$\lambda = \ell(\ell+1) \quad \ell = 0, 1, 2, 3 \dots$$

$$m = 0, \pm 1, \dots, \pm \ell$$

$$L^2 \psi_{\ell, m}(\theta, \varphi) = \ell(\ell+1) \hbar^2 \psi_{\ell, m}(\theta, \varphi)$$

$$L_z \psi_{\ell, m}(\varphi, \theta) = m \hbar \psi_{\ell, m}$$

$$m = 0, \pm 1, \pm 2, \dots, \pm \ell$$

$$\sqrt{l(l+1)\hbar^2}$$

$$\text{length} = \sqrt{6}\hbar$$

z projection

$$2\hbar, 1\hbar, 0\hbar, -1\hbar, -2\hbar$$

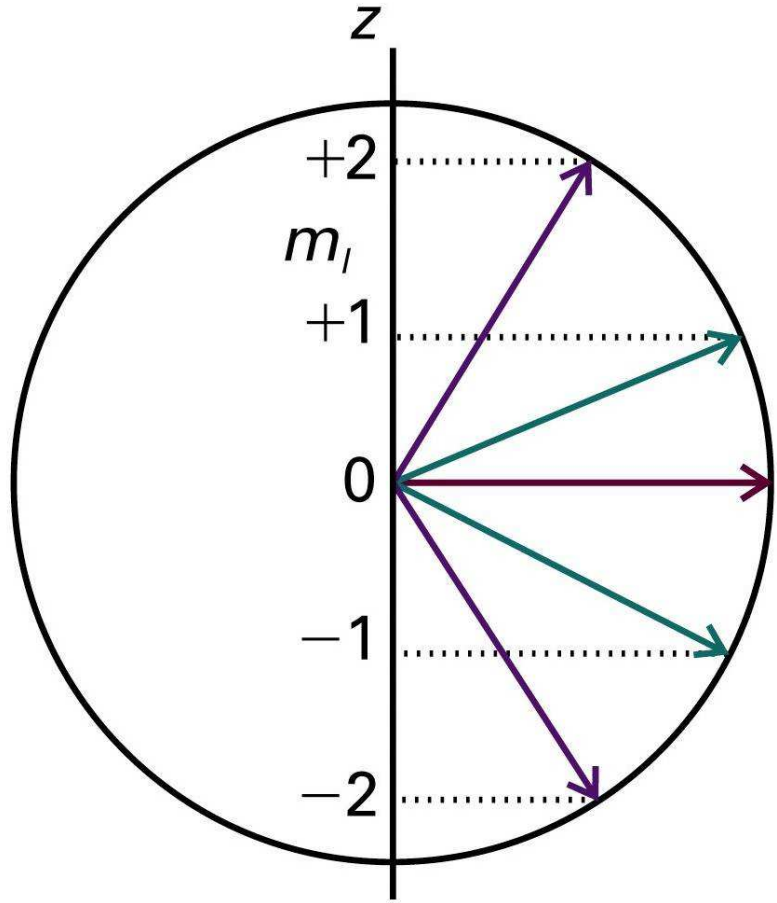


Figure 9-40a
Atkins Physical Chemistry, Eighth Edition
© 2006 Peter Atkins and Julio de Paula

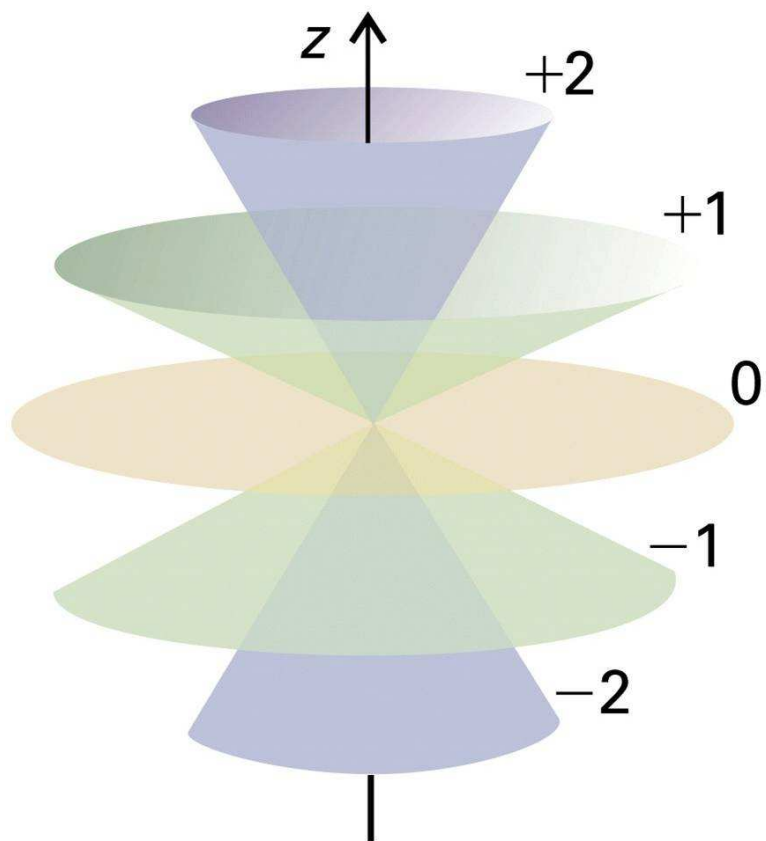
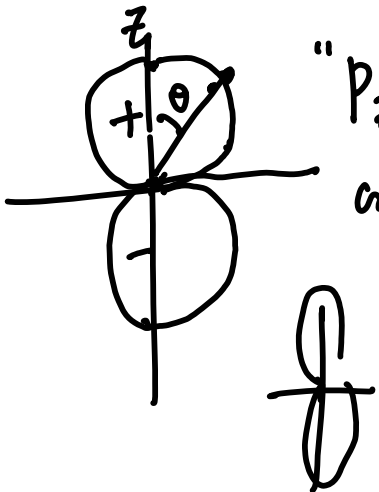


Figure 9-40b
Atkins Physical Chemistry, Eighth Edition
© 2006 Peter Atkins and Julio de Paula

Table 9.3 The spherical harmonics

l	m_l	$Y_{l,m_l}(\theta,\varphi)$
0	0	$\left(\frac{1}{4\pi}\right)^{1/2}$
1	0	$\left(\frac{3}{4\pi}\right)^{1/2} \cos \theta$
	± 1	$\mp \left(\frac{3}{8\pi}\right)^{1/2} \sin \theta e^{\pm i\varphi}$
2	0	$\left(\frac{5}{16\pi}\right)^{1/2} (3 \cos^2 \theta - 1)$
	± 1	$\mp \left(\frac{15}{8\pi}\right)^{1/2} \cos \theta \sin \theta e^{\pm i\varphi}$
	± 2	$\left(\frac{15}{32\pi}\right)^{1/2} \sin^2 \theta e^{\pm 2i\varphi}$
3	0	$\left(\frac{7}{16\pi}\right)^{1/2} (5 \cos^3 \theta - 3 \cos \theta)$
	± 1	$\mp \left(\frac{21}{64\pi}\right)^{1/2} (5 \cos^2 \theta - 1) \sin \theta e^{\pm i\varphi}$
	± 2	$\left(\frac{105}{32\pi}\right)^{1/2} \sin^2 \theta \cos \theta e^{\pm 2i\varphi}$
	± 3	$\mp \left(\frac{35}{64\pi}\right)^{1/2} \sin^3 \theta e^{\pm 3i\varphi}$

Table 9-3
Atkins Physical Chemistry, Eighth Edition
© 2006 Peter Atkins and Julio de Paula

$l = 1, m = 0$
 $Y_{lm}(\theta, \varphi) = Y_{10}$
 $= \cos \theta$ $\cos \theta = 1$


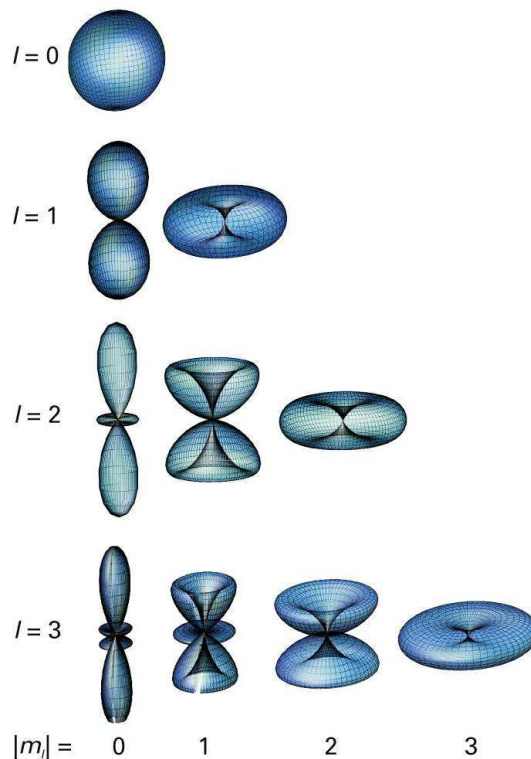


Figure 9-37
Atkins Physical Chemistry, Eighth Edition
 © 2006 Peter Atkins and Julio de Paula

p_x & p_y orbitals \rightarrow not eigenfunctions

of the \hat{L}_z operator

$$Y_{11} = \cos\theta e^{i\varphi}$$

$$Y_{1-1} = \cos\theta e^{-i\varphi}$$

$$p_x = Y_{11} + Y_{1-1}$$

$$p_y = Y_{11} - iY_{1-1}$$

Rotational spectroscopy

$$\hat{J}^2 = \frac{\hat{L}^2}{2I}$$

$$\hat{L}^2 Y_{lm}(\theta, \varphi) = l(l+1)\hbar^2 Y_{lm}$$

Rotational energies

$$E_l = \frac{l(l+1)}{2I} \hbar^2 \quad l = 0, 1, 2, 3, \dots$$

$$\text{--- } \frac{6\hbar^2}{2I}$$

$$\text{--- } \frac{2\hbar^2}{2I}$$

$$\text{--- } 0$$

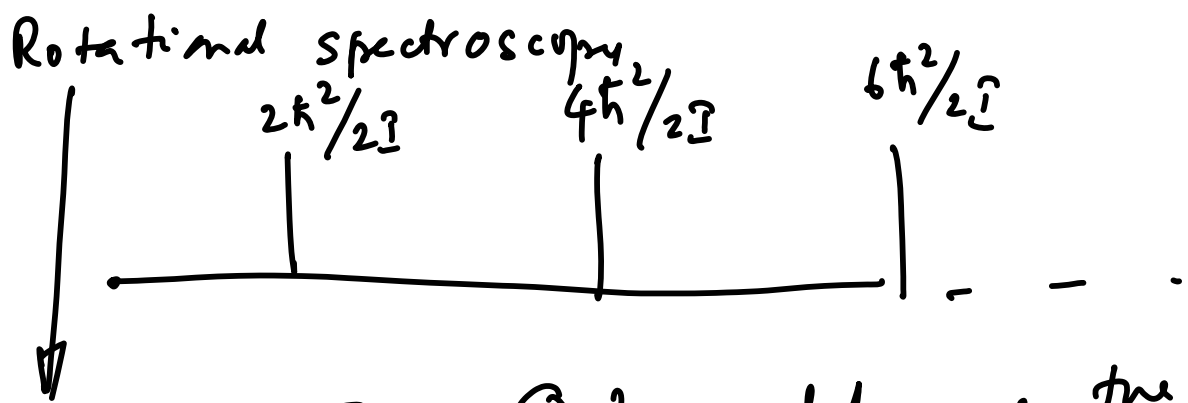
$$\Delta l = \pm 1$$

$$l=0 \rightarrow l=1$$

$$\Delta E = \frac{2\hbar^2}{2I}$$

$$l=1 \rightarrow l=2$$

$$\Delta E = \frac{4\hbar^2}{2I}$$



Microwave spectroscopy $I = \mu r^2$ determine the band length r

Vibration — Infrared
 Rotation — Microwave